PYRONE DERIVATIVES FROM *PODOLEPIS HIERACIOIDES* AND SESQUITERPENE ACIDS FROM *CASSINIA LONGIFOLIA*

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Abstract - The aerial parts of *Podolepis hieracioides* afforded eight γ -pyrones all derived from 2-methoxy-3,5-dimethyl-6-undecyl- γ -pyrone (podopyrone). The aerial parts of *Cassinia longifolia* gave in unusual high concentration two sesquiterpene acids, both related to costic acid. The structures were elucidated by highfield NMR spectroscopy. The structures of some previously reported pyrones are revised.

INTRODUCTION

The problems of the botanical relationships of the Australian members of the subtribe Gnaphaliinae (tribe Inuleae) have been discussed [1]. As little is known concerning their chemistry, we have started studying the chemistry of several of these genera. In this paper we report on the constituents of *Podolepis hieracioides* F. Muell. and Cassinia longifolia R. Br.

RESULTS AND DISCUSSION

The Australian genus Podolepis is placed in the subtribe Gnaphaliinae in a group together with parts of the genus Athrixia which is close to the Australian Helichrysum group [1]. So far none of the 18 species has been studied chemically. The aerial parts of P. hieracioides contain a complex mixture of compounds which by thin layer chromatography and reversed phase HPLC afforded nine compounds (1 8 and 11).

The ¹H NMR spectrum of 1 (see Experimental), molecular formula $C_{19}H_{32}O_3$, was not very conclusive. The presence of four methyl groups was indicated by singlets at δ 1.92, 1.82 and 3.93 and by a triplet at δ 0.86. The latter signal together with a broadened triplet, a triplet of triplets and a multiplet between $\delta 1.35-1.20$ (16H), showed the presence of a long chain saturated side chain. The IR band at 1670 cm⁻¹ and the ¹³C NMR spectrum (see Experimental) indicated the presence of a ypyrone ring system. However, the relative position of the substituents could not be deduced unambigiously from the data. A homoallylic coupling between the signals at $\delta 2.57$ and 1.92 showed that the aliphatic chain was in a position ortho to one of the methyl groups. Furthermore, clear NOEs were observed between H-7 and H-1' as well as between H-8 and OMe while no NOE was present between H-7 and OMe. In agreement with all data therefore the structure 1 was proposed. A gated decoupled 13C NMR spectrum confirmed the presence of a y-pyrone. The signal at 162.2 ppm was a quartet of quartets which by irradiation of the methoxy carbon was changed to a

quartet and therefore was due to C-1. This was also supported by the fragmentation pattern in the mass spectrum of 1. The base peak was at m/z 168 (C₉H₁₂O₃) and was probably formed by a McLafferty-like

11

188 C. ZDERO et al.

fragmentation of the side chain with the ring oxygen. Compound 1 we have named podopyrone.

The ¹H NMR spectrum and the molecular formula of 2 clearly showed that this pyrone was a nor-derivative of 1.

In the ¹H NMR spectrum of the main constituent (6), one methyl singlet in the spectrum of 1 was replaced by the signals of an ethyl group and the methyl triplet by a doublet at δ 1.18. The latter was coupled with a lowfield signal at δ 4.87. As an acetate methyl singlet was visible a 10'-acetoxy derivative was present. The position of the ethyl group followed from the ¹H NMR spectrum by the homoallylic couplings of H-1' with the methyl singlet. This was also supported by the ¹³C NMR data (see Experimental).

The data of compounds 3 and 7 were close to those of 6. However, the molecular formulae indicated that 3 had one methylene group less and 7 one more. The ¹H NMR spectra (see Experimental) showed that 3 was the 10'-acctoxy derivative of 1 and that 7 was the 7-methyl derivative of 6.

The ¹H NMR spectrum of 4 (see Experimental) differed from that of 3 by the absence of the methyl doublet. Spin decoupling indicated that a second two proton triplet was due to a methylene group in the side chain. The presence of a methyl ketone was deduced from the methyl singlet at $\delta 2.13$ and from the IR band at 1720 cm^{-1} . Accordingly, all the data agreed with structure 4 which was further supported by the fragmentation pattern in the mass spectrum ([M - COMe] and [M - CH₂COMe].

All the data of compound 5 indicated it was the desacetyl derivative of 6. As expected the H-10' signal was shifted upfield and the acetate methyl singlet was missing.

The molecular formula of 8 was identical with that of 3. However, the 1 H NMR spectrum clearly differed. The presence of an isomeric primary acetate followed from the two proton triplet at $\delta 4.05$. Thus compound 8 was 11'-acetoxypodopyrone.

The pyrones 3a and 5 7 are optically active. The absolute configuration was not determined. Comparison of the data of 1-8 with those of some methoxy pyrones from *Helichrysum* species indicated that these compounds are also γ -pyrones. Thus these structures have to be revised from α - to γ -pyrones (compounds 6-8 in [2], compounds 3-5 and 6 in [3] and compound 9 in [4]).

The lignane derivative 11 had been prepared by alanate reduction of the corresponding diester [5] and from a Geigeria species the dissobutyrate was reported [6].

The aerial parts of Cassinia longifolia gave large amounts of ledol and of γ - and δ -cadinene. In addition, mikanin and also in high concentration two isomeric costic acid derivatives (9a and 10a) were isolated. The structure of 9a, which was also transformed to the methyl ester 9b, followed from the ¹H NMR spectrum (Table 1). All signals could be assigned by spin decoupling. Most were close to those of costic acid. However, the typical H-7 signals were missing and the chemical shifts of H-13 were perturbed. Accordingly, a hydroxy group, its presence already deduced from the IR spectrum, was at C-7. The stereochemistry was deduced from the downfield shift of H-5 and was established by NOE difference spectroscopy. Clear effects were observed between the hydroxyl proton, H-5 and H-13', between H-5, OH and H-3 α as well as between H-14', H-2 β , H-6 β and H-8 β . The ¹³C NMR spectrum (see Experimental) also agreed with the proposed structure.

The ¹H NMR spectrum of 10b (Table 1), which was

Table 1. ¹H NMR spectral data of compounds 9a/b and 10a/b (400 MHz, CDCl₃, TMS as internal standard)

Н	9a	96	10a	106
lα	1.33 m	1.26 m	1.25 ddd	1.24 ddd
1 <i>β</i>	1.45 m	1.42 m	1.50 d(br)	1.47 b d (br)
2α 2β	}1.60 m	} 1.60 m	1.64 m	1.60 m
3 x	2.04 ddd	2.02 ddd (br)	1.97 ddd (br)	1.95 m
3β	2.31 dt	2.27 dt	2.33 dt	2.30 dt
5	2.37 dd (br)	2.36 dd (br)	1.89 d (br)	1.85 d (br)
6α 6β	}1.62 m	} 1.60 m	2.72 ddd 2.04 dd	2.65 dt 1.98 dd
8α	1.33 m	1.30 m	3.38 <i>dddd</i>	3.05 dddd
8β	1.75 m	1.75 m	2.16 <i>ddd</i>	2.13 ddd
9α	1.87 m	1.83 m	1.38 <i>ddd</i>	1.34 ddd
9₿	1.40 m	1.30 m	1.64 m	1.60m
13 13'	6.38 s 5.94 s	6.15 s 5.81 s	4.40 s (br)	4.33 s (br)
14	0.73 s	0.70 s	0.86 s	0.83 s
15	4.71 q	4.66 q	4.79 q	4.76 q
15'	4.38 q	4.33 q	4.48 q	4.46 q
ОМе	_	3.75 s		3.76 s

J[Hz]: Compounds 9a/b: 2α , $3\alpha = 7$; 2α , $3\beta = 12$; 2α , $3\beta = 2\beta$, $3\beta = 3$; 3α , $3\beta = 5$, $6\beta = 8\beta$, $9\alpha = 9\alpha$, $9\beta = 13$; 3, 15 = 5, 15 = 1; compounds 10a/b: 1α , $1\beta = 3\alpha$, $3\beta = 6\alpha$, $6\beta = 13$; 1α , $2\alpha = 6$; 3, 15 = 5, 15 = 1; 1α , $2\beta = 12$; 2α , $3\alpha = 7$; 2β , $3\alpha = 12$; 2α , $3\beta = 2\beta$, $3\beta = 3.5$; 5, $6\beta = 12$; 5, 6α , $8\alpha = 2$; 8α , $8\beta = 8\alpha$, $9\beta = 9\alpha$, $9\beta = 13$; 8β , $9\beta = 4.5$.

prepared from the naturally occurring acid, was in part similar to that of 9b. However, the presence of a primary alcohol followed from the broadened two proton singlets at $\delta 4.33$ and the signals of H-6 and H-8, especially that of H-8 α were shifted downfield. This indicated the configuration of the 7,11-double bond. The 13 C NMR spectrum (see Experimental) also agreed with the proposed structure. Thus 10a was the product of an allylic rearrangement of 9a and therefore 10a could be an artifact. This was excluded by the results of acid treatment of 9a. In addition to elimination of water only the Δ^3 isomer of 9a was obtained. The structure of which was easily deduced from the 1 H NMR spectrum (see Experimental).

The chemistry of the *Podolepis* species investigated indicated no relationship with the South African Athrixia species where diterpenes and thymol derivatives are common [7, 8]. It would be of interest to study the Australian Athrixia species which are placed in one group with *Podolepis*, unlike the South African Athrixia species which are members of another subtribe. As the rare substituted pyrones have been reported so far only from Helichrysum [2-4] and Gnaphalium [9 and unpublished results] the pyrones 1-8 may support the proposed close relationship of these genera [1].

The constituents from Cassinia, where so far only polyacetylenes were reported [10], show no clear relationship to other genera. Acids derived from eudesmane are widespread in the Compositae. However, the unusual high concentration of 9a and 10a in C. longifolia is remarkable. Clearly many more of the Australian species have to be investigated to get a better picture.

EXPERIMENTAL

The air dried plant material was collected in February 1986 near Canberra, Australia, and worked-up as reported previously [11]. CC of the defatted extract of the aerial parts of Cassinia longifolia (760 g, voucher Robinson 86-0064, all deposited in the U.S. National Herbarium) gave four fractions (1: petrol and Et₂O-petrol, 1:20, 2: Et₂O-petrol, 1:3, 3: Et₂O-petrol, 1:1 and 4: Et₂O). PTLC (silica gel, PF 254, Et₂O petrol, 1:9) of fraction 1 gave 1.6 g ledol and a mixture of hydrocarbons, one tenth of which was separated on AgNO3 coated silica gel (Et2O petrol, 1:100) affording 50 mg δ -cadinene (R_f 0.75) and 50 mg γ cadinene (R_f 0.50). PTLC of fraction 2 (Et₂O petrol, 1:3) gave 1 g ledol and 1 g 9a. Fraction 3 contained 7.6 g 9a. Fraction 4 gave by flash chromatography (silica gel ϕ 30-60 μ) 200 mg 9a, 2 g 10a and 300 mg mikanin. CC of the defatted extract of the aerial parts of Podolepis hieracioides (470 g, voucher Robinson, 86-0068) gave three fractions (1: petrol, 2: Et₂O-petrol, 1:1 and Et₂O, 3: Et₂O MeOH, 9:1). Fraction 1 gave by TLC (petrol) 20 mg squalene. HPLC of fraction 2 (RP 8, MeOH H₂O, 17:3, ca 100 bar) afforded 300 mg 6 (R_i 2.6 min), 55 mg 2 (R_i 3.9 min), 80 mg 1 (R, 4.9 min) and crude 7 which gave after repeated HPLC (same conditions) 20 mg 7 (R, 3.0 min). PTLC of fraction 3 (Et₂O-petrol, 3:1) gave two bands (3/1 and 3/2). HPLC of 3/1 (RP 8, MeOH H₂O, 4:1) gave 6 mg 4 (R_f 3.6 min), 16 mg 3 (R_f 5.0 min) and 1 mg 8 (R, 6.5 min). HPLC of 3/2 (same conditions) gave 3 mg 5 (R, 3.8 min) and crude 11 (R, 1.2 min) which gave by PTLC (Et₂O-petrol, 3:1, two developments, R_1 0.4) 40 mg 11.

Podopyrone (1). Colourless oil; IR $v_{max}^{CCl_a}$ cm⁻¹: 1670, 1610 (γ-pyrone); MS m/z (rel. int.): 308.235 [M]* (48) (calc. for $C_{19}H_{32}O_3$: 308.235), 293 [M - Me]* (35), 181 [$C_{10}H_{13}O_3$]* (70), 168.079 [$C_9H_{12}O_3$]* (100); ¹H NMR (CDCl₃): δ1.82 (s, H-7), 1.92 (s, H-8), 2.57 (t, H-1'), 1.63 (tt, H-2'), 1.35-1.20 (m, H-3'-H-10'), 0.86 (t, H-11'), 3.93 (s, OMe) (J [Hz]: 1',2' = 7.5; 10',11' = 6.5); ¹³C NMR (CDCl₃): $C_7 = C_7 = C_7$

nor-Podopyrone (2). Colourless oil; $1R v_{mai}^{CCL} \cdot cm^{-1}$: 1665, 1605 (7-pyrone); MS m/z (rel. int.): 294.220 $[M]^*$ (21) (calc. for $C_{18}H_{30}O_3$: 294.220), 279 $[M-Me]^*$ (28), 195 (25), 181 (52), 168 (100); 1H NMR as 1, except 0.87 (r, H-10').

10'-Acetoxypodopyrone (3). Colourless oil; IR $v_{\text{max}}^{\text{CCL}_4}$ cm⁻¹: 1735, 1240 (OAc), 1670, 1610 (γ -pyrone); MS m/z (rel. int.); 366.241 [M]* (32) (calc. for $C_{21}H_{34}O_5$: 366.241), 323 [M - COMe]* (62), 307 [M - OAc]* (28), 306 [M - HOAc]* (21), 195 (32), 181 (86), 168 (100); ¹H NMR (CDCl₃) as 1, except 4.87 (tq, H-10', J = 7 and 6 Hz), 1.19 (d, H-11'), 2.02 (s, OAc), 1.57 (m, H-9'); $[\alpha]_{D}^{2d} = -1$ $[\alpha]_{446}^{245} = -4$ (CHCl₃; c 1.6).

10'-Oxo-8-methylpodopyrone (4). Colourless oil; IR $v \le 1^{-2}$ cm⁻¹: 1720 (C=O), 1670, 1605 (7-pyrone); MS m/z (rel. int.): 336.230 [M]* (72) (calc. for $C_{20}H_{32}O_4$: 336.230), 321 [M - Me]* (100), 293 [321 - CO]* (12), 279 [M - CH₂COMe]* (21), 195 (30), 182 [$C_{10}H_{14}O_3$]* (31); ¹H NMR (CDCl₃) as 1, except 2.39 (q, H-8), 1.03 (t, H-9), 1.57 (m, H-8'), 2.41 (t, H-9'), 2.13 (t, H-11') (t [Hz]: 8,9 = 8',9' = 7).

10'-Hydroxy-8-methylpodopyrone (5). Colourless oil; $IR \, v_{\text{max}}^{\text{CCL}_4}$ cm⁻¹: 3640 (OH), 1665, 1605 (y-pyrone); MS m/z (rel. int.); 338.246 [M]* (41) (calc. for $C_{20}H_{34}O_4$: 338.246), 323 [M - Me]* (26), 305 [323 - H_2O]* (78), 195 (62), 182 (62), 55 (100), ¹H NMR (CDCl₃) as 1, except 2.39 (q, H-8), 1.03 (t, H-9), 1.60 (m, H-9'), 3.78 (tq, H-10', J = 6 and 6 Hz), 1.09 (d, H-11', J = 6 Hz), 10'-Acetoxy-8-methylpodopyrone (6). Colourless oil; $IR \, v_{\text{max}}^{\text{CCL}_3}$ cm⁻¹: 1740, 1240 (OAc), 1670, 1605 (y-pyrone); MS m/z

(rel. int.): 380.256 [M]* (64) (calc. for $C_{22}H_{36}O_5$: 380.256), 337 [M - COMe]* (26), 321 [M - OAc]* (24), 320 [M - HOAc]* (19), 305 [320 - Me]* (100), 195 (38), 182 (41); ¹H NMR (CDCl₃) as 3, except 2.38 (q, H-8) and 1.01 (t, H-9, J = 7 Hz); ¹³C NMR (CDCl₃, C-2-C-9): 162.0 s, 105.2 s, 180.3 s, 118.4 s, 158.2 s, 9.7 q, 25.2 t, 12.7 q; C-1' C-11': OMe: 55.1 q; OAc: 170.5 s, 21.2 q; C-1' C-11': 26.9, 28.9, 29.2, 29.3, 30.6, 29.2, 29.1, 15.1, 35.7 t, 70.8 d, 19.8 q.

10'-Acetoxy-7,8-dimethylpodopyrone (7). Colourless oil; $IR \nu_{\text{max}}^{\text{CCT}_4}$ cm⁻¹: 1740, 1240 (OAc), 1670, 1605 (7-pyrone); MS m/z (rel. int.): 394.272 [M]* (12) (calc. for $C_{23}H_{38}O_5$: 394.272), 335 [M - OAc]* (10), 334 [M - HOAc]* (9), 209 (100), 196 (40); ¹H NMR (CDCl₃) as 6, except 2.42 (q, H-7) and 1.05 (t, H-10, J = 7 Hz).

11'-Acetoxypodopyrone (8). Colourless oil; $IR \nu_{\text{max}}^{\text{CCI}_4} \text{ cm}^{-1}$: 1745, 1250 (OAc), 1670, 1610 (7-pyrone); MS m/z (rel. int.); 366.241 [M]* (28) (calc. for $C_{21}H_{34}O_5$: 366.241), 323 [M - COMe]* (27), 307 [M - OAc]* (33), 195 (36), 181 (74), 168 (100); 1H NMR (CDCI₃) as 1, except 4.05 (t, H-11', J = 6.5 Hz).

7-Hydroxyeudesm-4(15),11(13)-dien-12-acid (9a). Colourless crystals, mp 220°; 1R v_{max} cm ⁻¹: 3600 (OH), 3500-2600, 1700, 1650 (C=CCO₂H); $[\alpha]_{D}^{24} = +21$ (CHCl₃; c 3.03); Addition of CH₂N₂ in Et₂O gave the methyl ester 9b; colourless oil; IR $v_{\text{max}}^{\text{CCI}_{\bullet}}$ cm $^{-1}$: 3610 (0H), 1725, 1650 (C=CCO₂R); MS m/z (rel. int.): 264.173 [M]* (0.8) (calc. for C₁₆H₂₄O₃: 264.173), 246 [M -H₂O]* (100), 231 [246 - Me]* (76), 218 (52), 214 (29), 187 (68), 186 (62), 171 (62), 123 (98), 95 (62), 93 (68), 91 (62), 79 (64), 55 (66); 13C NMR (CDCl₃, C-1, C-15); 41.3 t, 23.4 t, 36.8 t, 150.3 s, 43.7 d, 34.3 t, 72.9 s, 31.3 t, 35.9 t, 35.4 s, 146.7 s, 167.8 s, 123.3 t, 15.4 q, 105.1 t; OMe: 51.7 q. 50 mg 9n in 5 ml dioxane were heated 4 hr with 5 ml 1M H₂SO₄ at 80°. After addition of CH₂N₂ to the crude reaction product PTLC (Et₂O-petrol, 1:3) gave a mixture of isomeric trienes and 20 mg of 9c. ¹H NMR (CDCl₃): 2.07 (dd (br), H-3a), 1.90 (d (br), H-3\beta), 2.72 (dd, H-6a), 2.37 (d (br), H-6\beta), 2.12 (dd, H-9a), 6.20 (d, H-13), 5.92 (d, H-13'), 1.08 (s, H-14), 1.84 (s (br), H-15) (J [Hz]: 2β , $3\alpha = 10$; 3α , $3\beta = 18$; 6α , $6\beta = 14.5$; 6α , 8α = 2.5; $8\beta,9\alpha = 9\alpha,9\beta = 13.5$; $8\alpha,9\alpha = 4$; $6\beta,13 = 0.7$.

13-Hydroxyeudesm-4(15),7(11)-dien-12-acid (10a). Colourless crystals, mp 156°; IR $v_{\text{CCL}}^{\text{CCL}}$ cm $^{-1}$: 3610 (OH), 3500-2600, 1695, 1650 (C=CCO₂H); MS m/z (rel. int.): 250.157 [M]* (3) (calc. for $C_{15}H_{22}O_3$: 250.157), 232 [M - H_2O]* (100), 217 [M - Me]* (56), 204 (42), 187 (36), 121 (66), 93 (80), 91 (82); $[\alpha]_{D}^{24}$ = +21 (CHCl₃; c 0.22). Addition of CH₂N₂ gave 10b, colourless oil; IR $v_{\text{CCL}}^{\text{CCL}}$ cm $^{-1}$: 3620 (OH), 1725, 1650 (C=CCO₂R); MS m/z (rel. int.): 264.173 [M]* (3) (calc. for $C_{16}H_{24}O_3$: 264.173), 246 [M - H_2O]* (100), 231 (66), 218 (32), 214 (26), 187 (58), 186 (48), 171 (52), 93 (82), 91 (81); ^{13}C NMR (CDCl₃, C-1 C-15); 41.0 t, 29.4 t, 41.8 t, 149.8 t, 50.6 t, 27.4 t, 154.0 t, 23.1 t, 36.4 t, 36.1 t, 125.1 t, 169.7 t, 59.0 t, 15.9 t, 105.8 t; OMe: 51.5 t.

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190 C. ZDERO et al.

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